

2,2,2-Trimethyl-*N*-(4-methylphenyl)acetamideB. Thimme Gowda,^{a*} Jozef Kožíšek,^b Miroslav Tokarčík^c and Hartmut Fuess^d^aDepartment of Chemistry, Mangalore University, Mangalagangotri 574 199, Mangalore, India, ^bDepartment of Physical Chemistry, Slovak University of Technology, Radlinského 9, SK-812 37 Bratislava, Slovak Republic, ^cDepartment of Chemical Physics, Slovak University of Technology, Radlinského 9, 812 37 Bratislava, Slovak Republic, and ^dInstitute of Materials Science, Darmstadt University of Technology, Petersenstrasse 23, D-64287 Darmstadt, Germany

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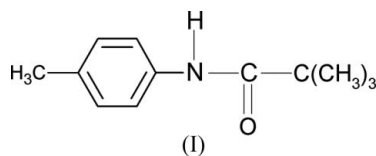
Key indicators

Single-crystal X-ray study
 $T = 299$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.073
 wR factor = 0.202
Data-to-parameter ratio = 15.8For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The structure of the title compound, $\text{C}_{12}\text{H}_{17}\text{NO}$, is closely related to the ring-unsubstituted 2,2,2-trimethyl-*N*-phenylacetamide, and *ortho*- and *meta*-ring-substituted amides, with somewhat slightly different bond parameters. $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonding results in the formation of an infinite chain running parallel to the *b* axis.

Comment

As part of a study directed at the systematization of the structures of *N*-aromatic amides (Gowda *et al.*, 2004, 2006, 2007*a,b*; Gowda, Kozisek, Svoboda & Fuess, 2007; Gowda, Paulus *et al.*, 2007), the structure of 2,2,2-trimethyl-*N*-(4-methylphenyl)acetamide (4MPTMA), (I), has been determined. The structure of 4MPTMA (Fig. 1) is closely related to the ring-unsubstituted 2,2,2-trimethyl-*N*-phenylacetamide (Gowda, Paulus *et al.*, 2007) and *ortho/meta*-ring-substituted amides (Gowda *et al.*, 2007*a,b*). The geometric parameters of 4MPTMA are similar to those of the corresponding *ortho*-methyl-substituted (2MPTMA) and *meta*-methyl-substituted (3MPTMA) amides. The dihedral angles between the benzene ring and the amide fragment in the three amides are 55.5 (2) (2MPTMA), 33.0 (1) (3MPTMA) and 32.8 (1)° (4MPTMA).



As observed in all other related structures, $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonding results in the formation of an infinite chain, running parallel to the *b* axis in this case (Table 1).

Experimental

The title compound was prepared according to the literature method of Gowda *et al.* (2003). The purity of the compound was checked by determining its melting point. It was characterized by recording its IR and NMR spectra (Gowda *et al.*, 2003). Single crystals of the title compound were obtained from an ethanol solution and used for X-ray diffraction studies at room temperature.

Crystal data

$\text{C}_{12}\text{H}_{17}\text{NO}$	$V = 2348.0$ (6) Å ³
$M_r = 191.27$	$Z = 8$
Orthorhombic, <i>Pbca</i>	Mo $K\alpha$ radiation
$a = 9.6900$ (12) Å	$\mu = 0.07$ mm ⁻¹
$b = 10.1287$ (16) Å	$T = 299$ (2) K
$c = 23.923$ (3) Å	$0.32 \times 0.19 \times 0.09$ mm

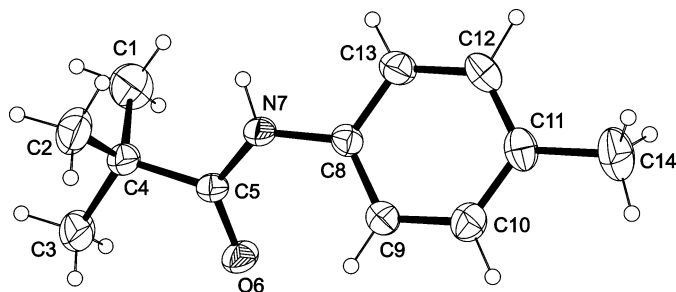


Figure 1
The molecular structure of the title compound, showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are represented as small spheres of arbitrary radius.

Data collection

Oxford Diffraction Xcalibur diffractometer
Absorption correction: none
14623 measured reflections

2304 independent reflections
1837 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.047$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.073$
 $wR(F^2) = 0.202$
 $S = 1.23$
2304 reflections
146 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.18 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.16 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N7-H7N\cdots O6^i$	0.87 (2)	2.12 (3)	2.993 (2)	176 (2)

Symmetry code: (i) $-x + \frac{1}{2}, y + \frac{1}{2}, z$.

All H atoms attached to C atoms were positioned geometrically and treated as riding, with $C-H = 0.93$ (CH aromatic) or 0.96 \AA (CH_3) and $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{CH})$ and $1.5 U_{\text{eq}}(\text{CH}_3)$. The position of the H atom attached to the N atom was refined freely; $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{N})$.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2006); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2006); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97* (Sheldrick, 1997), *PLATON* (Spek, 2003) and *WinGX* (Farrugia, 1999).

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